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Synthesis and SAR of novel isoquinoline CXCR4 antagonists with potent anti-HIV activity

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ABSTRACT

Using AMD070 as a starting point for structural modification, a novel series of isoquinoline CXCR4 antagonists was developed. A structure–activity scan of alternate lower heterocycles led to the 3-isoquinolinyl moiety as an attractive replacement for benzimidazole. Side chain optimization in the isoquinoline series led to a number of compounds with low nanomolar anti-HIV activities and promising rat PK properties.

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In recent years the medicinal chemist's focus in the battle against HIV/AIDS has broadened to include molecular targets beyond the HIV reverse transcriptase and protease enzymes, which are the cornerstones of the highly active antiretroviral therapy (HAART).¹ One area that has received a great deal of attention is the concept of blocking viral entry by targeting the chemokine receptors CCR5 and CXCR4, which function as co-receptors, along with CD4, to facilitate fusion of the viral membrane with the host cell.² CXCR4 is a G-protein coupled 7-transmembrane receptor utilized by T-tropic HIV strains to gain entry into T-cells. The appearance of CXCR4 utilizing strains of HIV is associated with a decrease in the number of T-cells and accelerated disease progression.³ In vitro studies have shown that addition of the natural CXCR4 ligand, SDF-1, or small molecule antagonists, can block HIV infection.^{2e} SDF-1 is a highly basic protein with about 20% of its 68 amino acids (for SDF- 1α) being arginine, lysine or histidine. This observation, along with the fact that the extracellular binding regions of the CXCR4 receptor are particularly rich in aspartic acid and glutamic

acid residues, points to a receptor–ligand binding model in which charge-charge interactions play a prominent role.⁴ Not surprisingly, the currently known small molecule CXCR4 antagonists are also highly basic in nature. A particularly notable example is the tetrahydroquinoline derivative AMD070 (Fig. 1) which was recently shown to possess significant anti-HIV efficacy in human clinical studies.⁵ Key features of the AMD070 pharmacophore include a triad of basic nitrogen atoms and a distal amino group attached to the central nitrogen by a 4-carbon tether.

We recently reported a structure–activity study toward improving the antiviral potency and/or ADME properties of AMD070 through iterative structural modifications. Specifically, we showed that it was possible to shift the distal amine side chain from the central nitrogen to either ring of the benzimidazole with retention of potent antiviral activity (Fig. 1, compounds 1 and 2).⁶ In a separate report we described related efforts in which we replaced the benzimidazole with an imidazopyridine in combination with the side chain shift.⁷ The results of these studies prompted us to explore the possibility of replacing the benzimidazole with other heteroaromatic ring systems capable of maintaining the required basic nitrogen triad.

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Figure 1. Tetrahydroquinolines with potent anti-HIV activity.

In order to identify viable alternative ring systems, we conducted a scan of various heteroaryl fragments with the distal amine side chain attached to the central nitrogen analogous to AMD070, with the exception that this study was done in the racemic series. Our synthetic approach is illustrated in Scheme 1. The key secondary amine intermediate **4** was synthesized by reductive amination of ketone **3**⁸ with *N*-Boc-1,4-diaminobutane. The benzimidazole derivative **5** was prepared via alkylation with *N*-Boc-2-(chloromethyl)benzimidazole followed by TFA deprotection. Compounds **6–14** were prepared by reductive alkylations with the appropriate heteroaryl aldehydes followed by HCl mediated cleavage of the Boc protecting group. The benzothiazole analog **15** was synthesized by alkylation with 2-(bromomethyl)-1,3-benzothiazole followed by acidic deprotection.

Antiviral and cytotoxicity data for the alternate lower heterocycle derivatives is shown in Table 1. Most of the ring systems studied retain appreciable antiviral activity with several in the same potency range as the benzimidazole derivative **5** (racemic version of AMD070). Exceptions include the 3-pyridyl and imidazole derivatives **7** and **12**, which due to the position of the basic ring nitrogens, are incapable of maintaining the required nitrogen triad

Scheme 1. Reagents and conditions: (a) *N*-Boc-1,4-diaminobutane, NaBH(OAc)₃, AcOH, 1,2-dichloroethane (73%); (b) *N*-Boc-2-(chloromethyl)benzimidazole, Kl, (*i*Pr)₂EtN, MeCN, rt (83%); (c) TFA, CH₂Cl₂ (71%); (d) RCHO, NaBH(OAc)₃, AcOH, 1,2-dichloroethane (55% for **6**); (e) 4 N HCl/dioxane, MeOH (62% for **6**); (f) 2-(bromomethyl)-1,3-benzothiazole, Kl, (*i*Pr)₂EtN, MeCN, 80 °C (59%).

Table 1 Anti-HIV $IC_{50}s \pm standard$ deviation (n) and $CC_{50}s$ for alternate lower heterocycle

Compound	R	IC ₅₀ ^a (nM)	CC ₅₀ ^b (nM)
5 °	H N N	23 ± 2 (4)	13,000
6 ^d		70 ± 8 (4)	11,000
7		>20,000 (1)	>20,000
8 ^d	N	34 ± 7 (2)	12,000
9		12 ± 1 (2)	13,000
10	N-	11 ± 1 (2)	8200
11 ^e	Me N	120 ± 3 (2)	>20,000
12	Me N	>20000 (2)	>20,000
13	N	42 ± 8 (2)	>20,000
14 ^d	S	340 ± 29 (2)	9700
15 ^d	S	200 ± 12 (2)	13,000

 $^{^{\}rm a}$ HOS cells expressing hCXCR4/hCCR5/hCD4/pHIV-LTR-luciferase, HIV-1, CXCR4 strain (IIIB). Compounds were tested for their ability to block infection of the HOS cell line. IC50 is the concentration at which 50% efficacy in the antiviral assay was observed. $^{\rm 9}$

- ^c Compound previously reported in Ref. 10.
- ^d Compound previously reported in Ref. 11.
- ^e Compound previously reported in Ref. 12.

geometry. The isoquinoline derivatives **8–10** were particularly interesting with antiviral activities in the 10–30 nM range. A comparison of these compounds with the pyridyl derivative **6**, shows that benzo ring fusion confers a consistent improvement in activity. Compounds **14** and **15** show reduced activities relative to the corresponding nitrogen analogs (i.e., **15** compared to **5**), perhaps due to the reduced basicity of the thiazole and benzothiazole ring nitrogens.

 $^{^{\}rm b}$ CC $_{\rm 50}$ is the concentration at which 50% cytotoxicity is observed in the HOS cell line.

Having identified several suitable replacements for the benzimidazole ring system, we turned our attention to shifting the distal amine side chain from the central nitrogen to the lower heterocycle. Based on a combination of attractive potency and synthetic tractability, we chose the isoquinoline system **9** for further study. We explored two different modes of attaching the amino group to the 4-position of the isoquinoline ring system: 1- and 3-carbon tethers (Fig. 2, general structure **A**).

Scheme 2 illustrates the synthesis of the 1-carbon primary amine 23. In order to assess the pharmacodynamic effect of the distal amine side chains, we also prepared the unsubstituted parent compound 19 for comparison. The bromomethyl intermediate 17 was prepared by NBS bromination of commercially available 3-methylisoguinoline **16**. Reaction of **17** with the enantiopure tetrahydroquinoline derivative **18**⁶ gave **19** in high yield. The S absolute configuration of 18 was chosen based on the clear activity preference observed in our previous studies.^{6,7} Substitution at the 4-position of the isoquinoline was achieved by NIS iodination to give intermediate 20, which was then subjected to Pd catalyzed cyanation followed by NBS bromination. Benzylic bromide 21 was reacted with 18 to afford nitrile 22 which was subjected to Raney nickel reduction to give primary amine 23. Scheme 3 details the subsequent side chain elaboration of 23 to give target compounds 24-29.

Our synthetic route to the 3-carbon series is shown in Scheme 4. Heck reaction of iodide $\bf 20$ with acrylonitrile followed by NBS bromination gave $\bf 30$ as an E/Z isomer mixture. Base promoted S_N2 substitution of bromide $\bf 30$ with $\bf 18$ gave $\bf 31$ in high yield. Sodium borohydride reduction of $\bf 31$ afforded nitrile $\bf 32$, which upon Raney nickel reduction gave $\bf 33$. Primary amine $\bf 33$ was then further elaborated as described in Scheme 4 to afford compounds $\bf 34-37$.

Table 2 shows SAR data for the 4-substituted isoquinoline series. Interestingly, the unsubstituted analog 19 shows significant activity even in the absence of a basic, distal amine side chain. Attaching the 1-carbon primary amine moiety (compound 23) affords a sixfold increase in activity relative to 19. Dimethylation of the amino group leads to a twofold loss in activity. A comparison of 23 and 24 with the longer chain amino derivatives 33 and 34, shows that the 3-carbon chain is favored, a result consistent with our previously reported chain length SAR in the benzimidazole series.⁶ The equipotent guanidine derivatives 25 and 35 showed impressive activities and the best cytotoxicity windows in the entire series. The equivalent activities of these 1- and 3-carbon analogs contrasts with the modest but consistent potency preference for the 3-carbon chain length that is generally observed. Perhaps this is due to the ability of the guanidine moiety to present two basic nitrogens in a 1,3-relationship thereby allowing the proximal nitrogen of 35 to mimic the distal nitrogen of 25. A surprising aspect of the SAR in this series is the unexpectedly potent activity of the non-basic,

Figure 2.

Scheme 2. Reagents and conditions: (a) NBS, AIBN, CCl₄, reflux (66%); (b) compound **18**, (iPr)₂EtN, DMF, rt (98%); (c) NIS, glacial AcOH, 80 °C (56%); (d) Zn(CN)₂, Pd(Ph₃P)₄, DMF, 120 °C (96%); (e) NBS, AIBN, CCl₄, reflux (62%); (f) compound **18**, (iPr)₂EtN, DMF, rt (92%); (g) H₂ (50 psi), Raney Ni, 2 M NH₃/MeOH (53%).

Scheme 3. Reagents and conditions: (a) 37% aqueous formaldehyde, NaBH(OAc)₃, AcOH, 1,2-dichloroethane (77%); (b) *N,N*-di-Boc-1*H*-pyrazole-1-carboxamidine, 1:1 THF/CH₂Cl₂ (68%); (c) TFA, CH₂Cl₂ (87%); (d) MeSO₂Cl, (iPr)₂EtN, CH₂Cl₂ (70%); (e) Boc-glycine, HATU, (iPr)₂EtN, MeCN (72%); (f) TFA, CH₂Cl₂ (89% for **30**); (g) 37% aqueous formaldehyde, NaBH(OAc)₃, AcOH, 1,2-dichloroethane (84% for **30**); (h) Boc-L-proline, HATU, (iPr)₂EtN, MeCN (63%); (i) Boc-D-proline, HATU, (iPr)₂EtN, MeCN (71%).

capped amine derivatives **26**, **36** and **37**. In fact, the 3-carbon sulfonamide **37** is nearly as active as the corresponding primary amine **33**. Similarly interesting is the activity of nitrile derivatives **22** and **32**. This contrasts sharply with our previous observations in the benzimidazole series, where non-basic, acylated amine side chains showed substantial reductions in activity. Within the set of lower heterocycles that we have explored, there is clearly something unique to the isoquinoline system that apparently facilitates side chain-receptor interactions through non-charged forces (i.e., hydrogen bond and/or dipole-dipole interactions).

The amide derivatives **27–29** allowed us to examine 4-atom tethers between the basic nitrogen and the ring system, albeit in a more conformationally rigid manner. The dimethylaminoglycine analog **27** was quite potent with a 12 nM IC₅₀. Proline derivatives **28** and **29** show that conformationally constraining the basic amine into a five-membered ring leads to fourfold

Me N
$$A,b$$
 A,b A,c A,c

Scheme 4. Reagents and conditions: (a) acrylonitrile, Pd(OAc)₂, Ph₃P, Et₃N, DMF, 100 °C (77%); (b) NBS, AIBN, CCl₄, reflux (57%); (c) compound **18**, (iPr)₂EtN, DMF, rt (98%); (d) NaBH₄, iPrOH, rt, 5 days (71%); (e) H₂ (1 atm), Raney Ni, 2 M NH₃/MeOH (70%); (f) 37% aqueous formaldehyde, NaBH(OAc)₃, 1,2-dichloroethane (81%); (g) N,N'-di-Boc-1H-pyrazole-1-carboxamidine, 1:1 THF/CH₂Cl₂ (71%); (h) TFA, CH₂Cl₂ (91%); (i) N,N-dimethylcarbamyl chloride, (iPr)₂EtN, CH₂Cl₂ (75%); (j) MeSO₂Cl, (iPr)₅EtN, CH₂Cl₂ (83%).

33

reduction in activity for the *S*-enantiomer **28** with the *R*-enantiomer **29** remaining unchanged relative to the open chain analog. The 3-carbon amine derivative **34** is the most active analog in the entire series with a 5 nM IC₅₀ and a 130-fold cytotoxicity window. In comparing the overall SAR of the isoquinoline series with our previously reported benzimidazole⁶ and imidazopyridine series,⁷ we were able to achieve similar antiviral activities, however, the isoquinolines show somewhat diminished cytotoxicity windows.

To further explore the therapeutic potential of the isoquinoline series, a representative set of compounds was chosen for pharmacokinetic analysis in rats. Table 3 illustrates the PK parameters associated with five selected analogs. All of the compounds showed acceptable half life and clearance values and with the exception of the guanidine derivative **25**, they also showed at least some level of systemic exposure. The dimethyl urea derivative **36** showed the best exposure with a 20% oral bioavailability, thus demonstrating the potential of this structural class in regards to lower species PK.

Using AMD070 as a starting point for structural modification, we identified several viable heterocyclic replacements for the benzimidazole ring system. The 3-isoquinolinyl moiety was chosen for further structural manipulation, specifically by using our previously successful approach of shifting the distal amine side chain to the lower heterocyclic ring system. This led to a number of compounds with low nanomolar antiviral activities, and to the surprising observation of potent activity with non-basic side chain derivatives. In addition, several analogs were shown to have acceptable rat PK properties. While we were able to achieve potency and pharmacokinetic profiles similar to our benzimidazole and imidazopyridine series, the isoquinolines suffered from generally less favorable cytotoxicity windows. However, this exercise effectively validates the concept of combining heterocyclic replacements with side chain shifts to generate a pharmacologically viable and structurally novel chemotype in the tetrahydroquinoline series.

Table 2 Anti-HIV $IC_{50}s \pm standard$ deviation (n) and $CC_{50}s$ for 4-substituted isoquinoline

	·		ъ.
Compound	R	IC_{50}^{a} (nM)	$CC_{50}^{b}(nM)$
19	H	160 ± 19 (2)	650
22	CN	72 ± 9 (2)	1300
23	NH ₂	27 ± 11 (2)	3100
24	N	48 ± 7 (5)	5900
25	NH NH ₂	20 ± 5 (4)	9000
26	$N-S-CH_3$	41 ± 9 (4)	2000
27	N N N	12 ± 4 (5)	480
28	O CH ₃	44 ± 9 (2)	540
29	O CH ₃	12 ± 5 (2)	690
32	CN	21 ± 2 (2)	950
33	NH ₂	12 ± 2 (2)	1000
34	N	5.0 ± 2.0 (5)	650
35	NH NH NH ₂	17 ± 3 (2)	6500
36	N N N	38 ± 12 (4)	750
37	$\begin{array}{c c} & O \\ N-S \\ H & O \\ O \end{array}$	16 ± 2 (4)	690

a,b See footnotes a and b from Table 1.

Table 3Pharmacokinetic parameters of representative isoquinoline CXCR4 antagonists in rats

Compound	$t_{1/2}$ (h)	Cl (mL/min/kg)	$V_{\rm d}$ (L/kg)	%F
25	12	16	9.0	0
27	1.5	11	1.5	10
34	2.9	6.1	2.9	6
36	3.1	16	1.7	20
37	5.8	9.5	1.4	4

Half life $(t_{1/2})$, clearance (CI) and volume of distribution (V_d) calculated following 1 mg/kg IV doses. Percent oral bioavailability (%F) calculated following solution doses of 3 mg/kg.

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